

# Thermochemical Equations

- **Thermochemical equations** are balanced chemical equations that include the physical states of all reactants and products and the energy change.
- $2\text{H}_2\text{O}(l) \rightarrow 2\text{H}_2(g) + \text{O}_2(g) \quad \Delta H = 572 \text{ kJ}$
- $\text{CH}_4(g) + 2\text{O}_2(g) \rightarrow \text{CO}_2(g) + 2\text{H}_2\text{O}(g)$   
 $\Delta H = -802 \text{ kJ}$

# Chemistry Thermochemical Equations Study Guide Answers

**Annelies Wilder-Smith**



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