Answers

Unit 1 answers

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Summary questions

1.1

1 (a) randomly arranged (b) close together, randomly arranged (c) close together, only vibrate

2 Solid has a definite volume; gas has no fixed volume / spreads out to fill container completely

solid does not flow; gas can flow / spreads out everywhere

solid can't be compressed; gas can be compressed

3 (a) particles move faster, distance unchanged
(b) particles get closer together, speed unchanged

1.2

1 (a) boiling / evaporation (b) melting
(c) condensing

2 energy absorbed; particles go from only vibrating in place to sliding over each other

3 volume decreases because particles slow down/ have less kinetic energy and hit the walls of the syringe with less force.

4 volume increases because while pressure outside the syringe is lower than inside, gas particles inside hit the walls and plunger of the syringe with more force than those outside and push the plunger outwards until atmospheric pressure = pressure inside syringe

1.3

1 (a) liquid because 970°C is higher than the melting point but lower than the boiling point

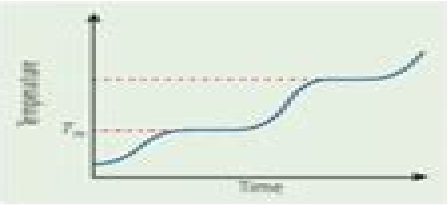
(b) gas because 1500°C is higher than the boiling point

2 (a) solid because −190°C is lower than the melting point

(b) gas because −150°C is higher than the boiling point

3 energy, melting, absorbed, forces, flat

4



T_m = melting point

5 energy absorbed is overcoming the forces holding the particles close together instead of raising the temperature

1.4

1 (a) mixture of solute in a solvent

(b) (compound) not containing water in its structure

(c) does not dissolve in a (particular) solvent

(d) solution in water

2 tap water contains impurities which may interfere with the experiment

3 $4 \times \frac{1000}{125} = 32$ g/dm³

4 blue to pink

1.5

1 particles; gases; random; different; mixed; diffusion

2 (a) particles in solid break free from each other / dissolve in water; particles gradually spread out from higher to lower concentration; by random movements / collisions

(b) particles in liquid break free from each other / particles escape into the air; particles gradually spread out from higher to lower concentration; by random movements / collisions

3 methane; nitrogen; oxygen; carbon dioxide

End of unit summary questions

1 (a) the gradual spreading out and mixing up of different particles by random movement

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