

Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a) anode _____
- b) cathode _____
- c) anode _____
- d) cathode _____
- e) half-cell reactions: Zn _____
- f) silver-silver chloride electrode _____
- g) cell reaction: Zn _____
- h) Zn _____
- i) Write overall cell reaction: $\text{Zn} + 2\text{Ag}^+ \rightarrow \text{Zn}^{2+} + 2\text{Ag}$ (10 marks)

Voltaic Cells Study Guide

Lingjun Ying



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