

CHEMISTRY Terminology

The ABCs of chemistry

A

- A:** Symbol for atomic mass number
- absolute zero:** Lowest temperature; 0 Kelvin
- Act:** Symbol for the element actinium, $Z = 89$
- accuracy:** The closeness of measurement to true value
- acetate anion:** $C_2H_3O_2^-$, charge -1
- acetic acid:** $C_2H_4O_2$
- acetone:** Dimethyl ketone; organic solvent
- acid-base indicator:** A chemical that changes color if the pH changes
- acid-base titration:** A method for determining acid or base concentrations
- acid definitions:** See Arrhenius acid, Brønsted-Lowry acid, Lewis acid
- acid ionization constant (K_a):** Equilibrium constant for acid dissociation
- acidic solution:** pH below 7
- actinide:** Element with $Z = 90$ or above; radioactive
- activation energy**



- (E_a): A process's energy barrier
- adhesion:** Attraction of unlike molecules or materials
- Ag:** Symbol for the element silver, $Z = 47$
- Al:** Symbol for the element aluminum, $Z = 13$

- alcohol:** Organic compound with -OH group; ROH
- aldehyde:** Carbonyl group bonded to hydrogen and an organic group
- alkali metal family:** Lithium, sodium, potassium, rubidium, cesium, and francium; column #1
- alkaline earth metal family:** Beryllium, magnesium, calcium, strontium, barium, and radium; column #2
- alkane:** Hydrocarbon; all C-C single bonds
- alkene:** Hydrocarbon; 1 or more C=C double bonds
- alkyne:** Hydrocarbon; 1 or more C≡C triple bonds
- allotropes:** Two or more forms of an element
- alloy:** Solution of 2 or more metals
- alpha (α):** Greek letter that denotes radioactive particle and various scientific variables
- alpha particle:** Helium nucleus; charge +2

- Am:** Symbol for the element americium, $Z = 95$
- amalgam:** Alloy of mercury with other metal
- amine:** Organic base; RRR'N; modified ammonia
- amino acid:** Compound with organic acid and organic base property; forms proteins and peptides
- ammonia:** NH_3 , base
- ammonium cation:** NH_4^+ , charge +1
- ammonium hydroxide:** NH_4OH
- amphoteric oxide:** Exhibits both acid and basic properties

- amu:** Atomic mass unit; $1/12$ mass of C-12
- anion:** Ion with a negative charge
- anode:** Electrode that supports oxidation
- antibonding MO:** MO is less stable than separate AOs
- AO:** Abbreviation for atomic orbital

- aq:** Aqueous; "dissolved in water"
- Ar:** Symbol for the element argon, $Z = 18$
- aromatic:** Organic compound with a benzene ring
- Arrhenius acid:** Produces hydronium ion in water solution
- Arrhenius base:** Yields hydroxide ion in water solution
- Arrhenius equation for rate constant (k):**

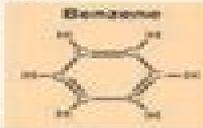
$$k = Ae^{-E_a/RT}$$

E_a = activation energy

- arsenide anion:** As^{3-} , charge -3
- As:** Symbol for the element arsenic, $Z = 33$
- At:** Symbol for the element astatine, $Z = 85$
- atm:** Symbol for pressure in "atmospheres"
- atom:** Fundamental unit of all matter
- atomic mass number (A):** Total number of protons and neutrons
- atomic number (Z):** Number of protons in the nucleus
- atomic orbital:** Wave motion of electrons in atoms
- atomic radius:** Empirical measure of atom size
- atomic weight:** Weighted average of natural isotopes of an element
- Au:** Symbol for the element gold, $Z = 79$
- Aufbau principle:** Guides the filling of electronic subshells of the elements
- Avogadro's law:** Volume is proportional to gas moles at fixed pressure and temperature
- Avogadro's number:** $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$

B

- B:** Symbol for the element boron, $Z = 5$
- Ba:** Symbol for the element barium, $Z = 56$
- balanced equation:** Each side of the equation has the same number of atoms of each element and charge
- balanced redox:** Electron loss = electron gain
- barium cation:** Ba^{2+} , charge +2
- base definition:** See Arrhenius base, Brønsted-Lowry base, Lewis base
- base ionization constant (K_b):** Equilibrium constant for base dissociation
- basic solution:** pH over 7
- battery:** Source of electrical power; galvanic cell
- Be:** Symbol for the element beryllium, $Z = 4$
- benzene:** C_6H_6 ; organic solvent; aromatic ring of 6 carbon atoms
- beryllium cation:** Be^{2+} , charge +2



- beta (β):** Greek letter that denotes radioactive particle and a member of scientific variables
- beta particle:** Energetic electron from the nucleus; charge -1
- Bh:** Symbol for the element bohrium, $Z = 107$
- Bi:** Symbol for the element bismuth, $Z = 83$
- bicarbonate anion:** HCO_3^- , charge -1
- bimolecular oxygen:** O_2
- binary compound:** Formed from 2 elements
- bisulfate anion:** HSO_4^- , charge -1

- Bk:** Symbol for the element berkelium, $Z = 97$
- bleach:** Chemical that is a strong oxidizing agent
- boiling:** Liquid \rightarrow gas at the boiling point
- boiling point elevation:** A solution has a higher boiling point than a pure solvent
- boiling point (T_b):** Liquid-gas equilibrium, $P = 1$ atm
- boiling point of water:** 100°C
- bond energy:** Energy held by a chemical bond
- bond length:** Distance between 2 atoms in a bond
- bond order for Lewis model:** # of bonds divided by # of bonded atoms
- bond order for MO treatment:** # of filled bonding MOs minus # of filled antibonding MOs
- bonding electrons:** Form chemical bonds in a compound
- bonding MO:** MO is more stable than AOs
- Boyle's law:** $PV = \text{constant for gas, fixed } T \text{ and } n$



- Br:** Symbol for the element bromine, $Z = 35$
- bromide anion:** Br^- , charge -1
- Brønsted-Lowry acid:** Proton donor
- Brønsted-Lowry base:** Proton acceptor
- buffer:** Composed of weak acid and weak base; serves to keep pH constant

C

- c:** Symbol for centi-, the SI prefix for $1/100$ or 10^{-2}
- c:** Symbol for the speed of light
- °C:** Celsius temperature scale
- C:** Symbol for the element carbon, $Z = 6$
- $C_2H_4O_2$: Acetic acid
- Ca:** Symbol for the element calcium, $Z = 20$
- calcium cation:** Ca^{2+} , charge +2
- calcium fluoride:** CaF_2
- calcium hydroxide:** $Ca(OH)_2$
- calorie:** English unit of heat energy
- calorimetry:** Study of the release or absorption of heat
- carbide anion:** C^{4-} , charge -4
- carbohydrate:** Organic compound; carbon bonded to several -OH groups (sugar and starch)
- carbon dioxide:** CO_2
- carbon monoxide:** CO
- carbonate anion:** CO_3^{2-} , charge -2
- carbonic acid:** H_2CO_3
- carbonyl group:** CO , ligand complex
- carboxyl group:** $>CO$ bonded to 2 groups
- carboxylic acid:** Organic acid; $R-COOH$
- catalyst:** Accelerates a reaction but is not consumed in the reaction; lowers the activation energy
- cathode:** Electrode that supports reduction
- cation:** Ion with a positive charge
- Cd:** Symbol for the element cadmium, $Z = 48$



College Chemistry 2 Study Guide

Daniela Niemeyer



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