

Dry Lab 3 Report Sheet

Atomic and Molecular Structure

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. The Mercury Spectrum

Instructor's approval of the wavelength grid of the spectra on the color plate (back cover): _____

B. The Spectra of Elements

1. Spectrum number ~~400~~¹⁰⁹ is the emission line spectrum for hydrogen on the color plate.

What are the wavelengths and colors of the emission lines of the visible spectrum of hydrogen?

Violet : 435 nm
 Blue-violet : 483 nm
 Blue-green : 484.9 nm
 Red : 647.1 nm

2. Identification of Spectra

- a. Spectrum number _____

Intense lines in the spectrum: _____ nm, _____ nm, _____ nm, _____ nm, _____ nm, _____ nm, _____ nm

Element producing the spectrum: _____

- b. Spectrum number _____

Intense lines in the spectrum: _____ nm, _____ nm, _____ nm, _____ nm, _____ nm, _____ nm, _____ nm

Element producing the spectrum: _____

C. Match of Molecule with Infrared Spectrum

Complete the following table.

<u>Molecule Assigned</u>	<u>Atom-in-Bond Arrangements</u>	<u>Absorption Bands (cm⁻¹)</u>	<u>Infrared Spectrum Figure No.</u>
_____	_____	_____	_____
_____	_____	_____	_____
_____	_____	_____	_____

Dry Lab 3 Atomic And Molecular Structure

R Barnett



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