AP CHEMISTRY

CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE & PERIODIC TRENDS– KEY

1)When a solution of sodium chloride is vaporized in a flame, the color of the flame is:
A) Blue **B) Yellow** C) Green D) Violet E) White

2) Electromagnetic radiation travels through vacuum at a speed of _____ m/s.
A) 186,000 **C) 3.00×10^{8}** E) It depends on λ
B) 125 D) 10,000

3) The wavelength of light that has a frequency of $1.20\times10^{13}\,s^{-1}$ is _____ m.
A) 25.0 **B) 2.50×10^{-5}** C) 0.0400 E) 2.5
D) 12.0

GIVEN: $\nu = 1.20 \times 10^{13}\ s^{-1}$ UNKNOWN: λ
CONNECTION: $c = \lambda\nu$
SUBSTITUTE & SOLVE: $\lambda = c/\nu$

$$\lambda = \frac{3.00 \times 10^{8}\ m \bullet s^{-1}}{1.20 \times 10^{13}\ s^{-1}}$$

$\lambda = 2.50 \times 10^{-5}$ m

4) Ham radio operators often broadcast on the 6-meter band. The frequency of this EM radiation is _____ MHz.
A) 500 B) 200 **C) 50** D) 20 E) 2.0
GIVEN: $\lambda = 6$ m UNKNOWN: ν *(Must convert answer to MHz!)*
CONNECTION: $c = \lambda\nu$
SUBSTITUTE & SOLVE: $\nu = c/\lambda$

$$\nu = \frac{3.00 \times 10^{8}\ m \bullet s^{-1}}{6\ m}$$

$$\lambda = 0.5 \times 10^{8}\ Hz \times \frac{1\ MHz}{10^{6}\ Hz} = 50\ MHz$$

5) The energy of a photon of light is _____ proportional to its frequency and _____ proportional to its wavelength.
A) directly, directly C) inversely, directly E) indirectly, not
B) inversely, inversely **D) directly, inversely**

6) The energy of a photon that has a wavelength of 12.3 nm is _____ J.
A) 1.51×10^{-17} B) 4.42×10^{-23} C) 1.99×10^{-25} D) 2.72×10^{-30} **E) 1.62×10^{-17}**

7) The wavelength of a photon that has an energy of 5.25×10^{-19} J is _____ m.
A) 3.79×10^{-7} B) 2.64×10^{6} C) 2.38×10^{23} D) 4.21×10^{-24} E) 3.79×10^{7}

8) A mole of red photons of wavelength 725 nm has _____ kJ of energy.
A) 2.74×10^{-19} B) 4.56×10^{-46} C) 6.05×10^{-3} **D) 165** E) 227

GIVENS: $\lambda = 725$nm (must convert to m before using formulas)

$$\lambda = 725\ nm \times \frac{1\ m}{10^{9}\ nm} = 7.25 \times 10^{-7}\ m$$

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