

1. Determine period and group from electron configuration below!

- $1s^2 2s^2 2p^2$
- $1s^2 2s^2 2p^4$
- $1s^2 2s^2 2p^6 3s^2 3p^2$
- $1s^2 2s^2 2p^6 3s^2 3p^4$
- $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2$

2. Determine the number of atom from the period and group below!

- Group II, period 2
- Group III, period 2
- Group II, period 4
- Group IV, period 3
- Group III, period 4

3. Look the figure below!



The right statement is ...

- No. 1 is proton
- No. 2 is proton
- No. 3 is neutron
- No. 2 is electron

4. Look the figure below!



The maximum electron of $n=2$ is ...

- 2
- 8
- 18
- 32

5. Configuration of atom is $1s^2 2s^2 2p^4 3s^2 3p^4 4s^2 3d^4$. Periodic and group of atom are ...

- Periodic 3 and group IV A
- Periodic 3 and group VI B
- Periodic 4 and group VI A
- Periodic 4 and group VI B

6. Atom ${}_Z X$ have shells are ...

- K, L, M, N
- K, L, M
- K, L
- K

7. Look at the picture!



The atom has charged.

- Netral
- Positif
- Negatif
- no

8. Configuration of atom is $1s^2 2s^2 2p^4 3s^2 3p^4 4s^1$. Periodic and group of atom are ...

- Periodic 3 and group I A
- Periodic 3 and group I B
- Periodic 4 and group I A
- Periodic 4 and group I B

9. Look the figure below!



The maximum electron of $n=3$ is

- 2
- 8
- 18
- 32

10. Look at the picture!



The atom has charged.

- Netral
- Positif
- Negatif
- no

11. The figure on the right shows the atomic shell of an atom.



The maximum numbers of electrons that can be occupied by the four shell are ...

- 2
- 8
- 18
- 32

12. Study the table on the right.

Number	The name of element
1	Chlorine
2	Aurum
3	Carbon
4	Fluorine

The elements that are not classified into halogen group are ...

- 1, 2 and 3
- 2 and 3
- 1 and 3
- 2 and 4

13. The electron configuration of an element is: 2, 8, 8, 2.

- The number of valence electron is 2
- The atom is included in alkali metal
- The period of the atom is 4
- The atom is include in earth alkali

The correct statement according to the configuration above are numbers....

- 1, 2 and 3
- 1, 2 and 4
- 1, 3 and 4
- 2, 3 and 4

14. The electron configuration of the unknown atom is 2, 8, 8, 2. The position of the atom in a periodic table is ...

- group III and period 3
- group II and period 4
- group IV and period 4
- group IV and period 3

15. The position of an atom in a periodic table is group II and period 4. The atomic number of the atom is ...

- 30
- 27
- 20
- 15

16. There are atoms belong to alkali metal group, **excepted** ...

- lithium
- sodium
- hydrogen
- potassium

Electron Configuration Test And Answer

Judd E. Hollander



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Whispering the Techniques of Language: An Emotional Quest through **Electron Configuration Test And Answer**

In a digitally-driven world wherever displays reign great and immediate communication drowns out the subtleties of language, the profound strategies and psychological nuances hidden within phrases often get unheard. However, nestled within the pages of **Electron Configuration Test And Answer** a fascinating fictional value sporting with organic thoughts, lies a fantastic journey waiting to be undertaken. Published by a talented wordsmith, this marvelous opus attracts visitors on an introspective journey, delicately unraveling the veiled truths and profound impact resonating within the very material of every word. Within the mental depths of this touching review, we shall embark upon a genuine exploration of the book's core themes, dissect its interesting publishing fashion, and yield to the strong resonance it evokes serious within the recesses of readers' hearts.

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Table of Contents Electron Configuration Test And Answer

1. Understanding the eBook Electron Configuration Test And Answer
 - The Rise of Digital Reading Electron Configuration Test And Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Electron Configuration Test And Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Electron Configuration Test And Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Electron Configuration Test And Answer

- Personalized Recommendations
 - Electron Configuration Test And Answer User Reviews and Ratings
 - Electron Configuration Test And Answer and Bestseller Lists
5. Accessing Electron Configuration Test And Answer Free and Paid eBooks
 - Electron Configuration Test And Answer Public Domain eBooks
 - Electron Configuration Test And Answer eBook Subscription Services
 - Electron Configuration Test And Answer Budget-Friendly Options
 6. Navigating Electron Configuration Test And Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Electron Configuration Test And Answer Compatibility with Devices
 - Electron Configuration Test And Answer Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Electron Configuration Test And Answer
 - Highlighting and Note-Taking Electron Configuration Test And Answer
 - Interactive Elements Electron Configuration Test And Answer
 8. Staying Engaged with Electron Configuration Test And Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Electron Configuration Test And Answer
 9. Balancing eBooks and Physical Books Electron Configuration Test And Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Electron Configuration Test And Answer
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Electron Configuration Test And Answer
 - Setting Reading Goals Electron Configuration Test And Answer
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Electron Configuration Test And Answer

- Fact-Checking eBook Content of Electron Configuration Test And Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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