

1. Write the equations depicting the behavior of the following acids and bases in water.

- $\text{HBr} \rightarrow \text{H}^+ + \text{Br}^-$
- $\text{HCN} \rightarrow \text{H}^+ + \text{CN}^-$
- $\text{RbOH} \rightarrow \text{Rb}^+ + \text{OH}^-$
- $\text{KOH} \rightarrow \text{K}^+ + \text{OH}^-$

2. Classify each reactant and product as an acid (A), base (B), conjugate acid (CA), or conjugate base (CB).

- $\text{F}^- + \text{H}_2\text{O} \rightarrow \text{HF} + \text{OH}^-$
- $\text{HClO} + \text{H}_2\text{O} \rightarrow \text{H}_3\text{O}^+ + \text{ClO}^-$
- $\text{H}_3\text{PO}_4 + \text{NH}_3 \rightarrow \text{NH}_4^+ + \text{H}_2\text{PO}_4^-$
- $\text{HNO}_2 + \text{HS}^- \rightarrow \text{H}_2\text{S} + \text{NO}_2^-$

a- base, acid, conjugate acid, conjugate base
 b- acid, base, conjugate acid, conjugate base
 c- acid, base, conjugate acid, conjugate base
 d- acid, base, conjugate acid, conjugate base

3. Classify each of the following acids as monoprotic, diprotic, triprotic, or tetraprotic.

- H_2SO_4 (sulfuric acid) diprotic

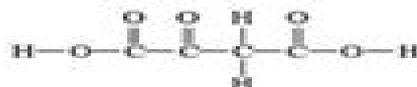
- Acetic acid, show to the right



monoprotic

- HCN (cyanic acid) monoprotic

4. Oxaloacetic acid, which is produced in metabolic reactions in the human body has the following structure.



- Classify the above structure as a mono-, di-, tri-, or tetraprotic acid. Explain your answer.

Diprotic acid because it can only transfer 2 hydrogen protons

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Richard Bailey



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